small animal - is readable on the prior art host. Only amendment of a distinctive host not readable on the prior art will obviate this rejection." The Chemical Abstracts reference cited in the rejection refers only to the rat as an example of a small animal. This is true also of British Specification 2101114 (copy enclosed) from which the Abstract derived.

In these circumstances, to draw an absolutely clear line of distinction between the present invention and the prior art, new claim 5 now specifies that the small mammal is a dog or cat (cf. claim 2). Claims 2 and 4 have been cancelled as redundant and a consequential amendment has been made in claim 3. It is believed, in these circumstances, that the amended claims are clearly distinguished from the prior art and relate to a utility which the Examiner agrees was not taught in the art.

Favorable reconsideration and allowance of the application with withdrawal of the final rejection are therefore believed to be in order and are earnestly requested.

Should minor issues remain which can be resolved by means of a telephone interview, the Examiner is respectfully requested to contact the undersigned attorney at the below-listed telephone number in order to resolve said issues.

In the event that this paper is not considered to be timely filed, applicants hereby petition for an appropriate extension of time. The fee for any such extension may be

charged to our Deposit Account No. 01-2395, along with any other required fees.

Respectfully submitted,

ARMSTRONG, NIKAIDO, MARMELSTEIN & KUBOVCIK

Ronald . Kubovcik Reg. No. 25,401

Atty. Case No. P418-566-K851068 Suite 912 1725 K Street, N.W. Washington, D.C. 20006 659-2930 RJK/ds

Attachment: Copy of British Specification 2101114

- (21) Application No 0121333
- (22) Date of filing 19 Jul 1961
- (43) Application published
- (51) BYT CL* C09D 233:56 233:64 405:08
- (52) Domestic classification C2C 1410 1474 200 204 213 215 220 226 227 227 246 247 250 252 253 267 28X 29X 29Y 204 305 307 313 314 31Y 337 336 339 351 353 35Y 260 362 263 36Y 386 43X 440 47X 569 50Y 623 625 652 655 639 770 776 777 778 77K 602 86Y AA UL UN UP WK 20 21
 - U1S 1311 1320 C2C
- (56) Decuments exted
- (58) Field of search
- (71) Applicants
 Farmos Group Ltd.,
 (Finland),
 PL 425,
 GF-20101 Yurku 10,
 Finland.
- (72) Inventors
 Arto Johannes
 Karjalainen,
 Esko Kelervo Pohjala,
 Kauko Olva Antero
 Kurkela.
- (74) Agents
 J. A. Kemp and Co.,
 14 South Square,
 Qray's Inn,
 London WC1R SEU.

- (64) Substituted imidazole derivatives and their preparation and use
- (57) Mainly nevel compounds of the formula

wherein R, is H, an ellest of I to 4 corden atoms of OH, OH, R, OH of OH, R, is

and Ra is H or OH; or Ra and Ra together represent CH2. CM CM.

R_b, R₀ and R₁, which can be the same or different, are H. CM₅ CM₅CM₅ hategon.

CH or - OCH₅ or R₀ is hydrogen and R₀ and R₁ tegether form an

-O - CH₅ - O-bridge between two adjacent carbon stems in the gheat group,

 $-O-CH_2-O$ -bridge between two adjacent carbon atoms in the phony group, $-CHR_8-$ is $-CH_2$. $-CH(CH_3)$. $-CH(-CH_2CH_3)$. $-CH(-CH_2CH_3)$.

CH₃
- CH₂CH₂CH₃, - CH₋CH₃ or - CH₂CH₂CH₃; n is 0 to 4; and their non-toxic pharmaceutically acceptable acid addition salts exhibit antihypertensive,

G

હ્ય

 \mathbb{C}_{3}

 Ξ

The present invention relates to substituted imidazole derivatives and their non-toxic, phermaceutically acceptable acid addition saits, and their preparation, to phermaceutical compositions containing the same, 5 and to their use.

The imidatole derivatives of the present invention have the general formula

20 wherein R, is M, an alkyl of 1 to 4 cerbon sterms or - Onyon; R, a M or Ony; R, a

 R_4 is H or OH; or R_3 and R_4 together represent =CH₂, =CH-CH₂, =CH-CH₂,

10

 R_8 , R_8 and R_7 , which can be the same or different are H. – CH_3 . – CH_3CH_4 hategen. CH or – CCH_3 or R_3 is hydrogen and R_8 and R_7 together form an – O – CH_7 – O – bridge between two adjacent carbon etams in the phenyl group;

45
$$-CHR_0 - is -CH_2 - , -CH(CH_2) - , -CH(-CH_7CH_3) - ,$$

GB 2 101 114 A 2 then R_s, R_s and R₇ are not all simultaneously hydrogen .. when R₁, R₂ and R₄ are all hydrogen and R₃ to 5 then R. R. R, are not all simultaneously hydrogen. .. Re and Re are not simultaneously hydrogen, and - R₁₁ and R₁₀ are not simultaneously hydrogen. 10 Because of the Lautomeriem in the imidezoic ring the compounds of the general forms 4(5)-substituted imidazole derivativ The non-toxic phermaceutically acceptable acid addition selts of these compounds are also avail scope of the invention. The compounds of the formula (I) and (II) form acid addition salts with both organic an They can thus form many phermacoutically usable acid addition salts, es, for instance, chloridge, brain sulfates, nitrates, phosphates, sulfonates, formates, tartrates, maleates, citrates, benveutes, saticylanes, accordates and the like. The invention includes within its scope pharmaceutical compositions comprising at reast one of the 20 compounds of formula (I) or (II) or a nontoxic, pharmaceutically acceptable saft thereof, and a companie pharmaceutically acceptable carrier therefor. The invention provides, for example, the following specific compounds of formula (I) 4-{a_a-bis(2-methylphenyl)hydroxymethyl)mxdazole 4-{[a-(2-methylphenyl)]-2-methylbenzyl $\}$ imidazole 25 4-(a-phenylbenzyl)-5-methylimidazole $4-[[\alpha-(2,6-dimethylphonyl)]-cr-methyl]hydroxymethyl]imidazolo$ 4-[[a-(2,3-dimethylphonyl)]-a-methyl]hydroxymethyl]imidazole 4- $\{\alpha,\alpha$ -bis(2-methylphenyl)hydroxymethyl)-5-methylimidazole 4-[[a-(2-methylphenyl)]-2-methylbenzyl]-5-methylimidazole 30 4-{(a-methyl)-2,6-dimethylbenzyl}imidazole 4-[(a-methyl)-2,3-dimethylbenzyl]imidazole 4-((a-ethyl)-3-methylbenzyl}imidazola 4-[(n-butyl)-2,3-dimethylbenzyl]imidazole 4-[(a-methyl)-2,3-dimethylbenzyl]-2-methylimidazole 35 4-{(a-propyl)-2-methylbenzyl]imidazole 4-((a-methyl)-2-methylbenzyl)imidazole 4-{(a-methyl)-2,5-dimethylbenzyl|imidazole 4-(a-ethyl-a-(3-methylphenyl)-hydroxymethyl]imidazolo 4-[α-butyl-α-(2,3-dimethylphenyl)-hydroxymethyl]imidazole 40 4-[α-methyl-α-(2,3-dimethylphenyl)-hydroxymethyl]-2-methylimidazole 4-(a-propyl-a-(2-methylphenyl)-hydroxymethyl]imidarole 4- $\{a$ -methyl-a- $\{2$ -methylphenyl}-hydroxymethyl $\}$ imidazole 4-[lpha-methyl-lpha-(2,5-dimethylphenyl)-hydroxymethyl]imidezole 4-[a,a-bis-(2,3-dimethylphenyl)hydroxymethyl]imidazole 45 4-[u-(2,3-dimethylphenyl)-2,3-dimethylbenzyl]imidazole 4-[(u-ethyl)-2,6-dimethylbenzyl]imidazole 4-[(a-ethyl)-2,3-dimethylbenzyl]imidazole 1-(4-imidazolyl)-1-(2,3-dimethylphenyl)ethylene 1-(4-imidazolyl)-1-(2,6-dimethylphenyl)ethylene The following specific compounds of formula (II): 4-[2-(2,6-dimethylphenyl)-1-methylethyl]imidazole 4-[2-(2,6-dimethylphenyl)propyl]imidazole 4-[2-(2,6-dimethylphenyl)-1-methylpropyl]imidazole 4-[2-(2,6-dimethylphenyl)-2-hydroxyethyl]imidazole 55 4-(2-phenylpropyl)imidazole 4-[2-(2,6-dimethylphenyl)-1-methylethenyl]imidezole 4-[2-(2,6-dimethylphenyl)-1-propenyl]imidazole 4-(2-methyl-4-phenyl-1-butenyl)imidazole 4-[2-(4-chlorophenyl)-1-methylpropyl]imidazole 60 4-[5-(2,6-dimethylphenyi)-1-methyl-1-pentenyl]imidezole

4-[3-(2,6-dimethylphenyl)-2-methyl-1-propenyl]imidazole 4-[2-(2,6-dimethylphenyl)-1-ethylethenyl]imidazole

CO 2 101 114 A 3 4-(2-(2,6-dimethylphenyl)-1-methylethenyl)-8-methylimidezole 4-(2-(2,6-dichlorophenyl)-1-methylvenenyl)midezole 4-(5-(2,6-dimethylphenyl)-1-methyl-1-pentenyl)imidezole 4-(3-(2,6-dimethylphenyl)-1-ethyl-1-propenyl)imidezole 5 4-(5-(2,6-dimethylphenyl)-1-methyl-1-pentenyl)-8-methylimiderole 4-(5-(2,6-dimethylphonyl)-1-methylpentyl)imidesale 4-(4-(2.6-dichlorophenyl)-1-methyl-1-butenyl)imidasole 4-(2-(2,6-dimethylphenyl)-1-ethylethyl)imidezole 4-[2-(2,4-dimethylphenyl)-2-ethylethyl)tmidszole 38 10 4-(2-(3,4-methylenedioxyphenyl)propyl)imidezole The compounds of the present invention have been found to pos Preliminary tests have shown that they also possess other valuable pharmacelogical gro example, antithrombotic effect. Antimycotic and antifungal properties have also b naistly satisfy the ch While all of the compounds of formula (1) and (fil) esse 15 certain groups of compounds remain preferred. One such preferred group is reprewherein R_a is hydrogen, R_b is alkyl and R_b , R_b and R_b , which can be the same or different, each are hydrogen. methyl, ethyl or halogen. Another preferred group of compounds is represented by farm, its (III), self-Re and Ry, which can be the same or different, each are hydrogen, methyl, ethyl or hategen. Especially the compounds wherein n is greater than 0 possess valuable antimycosic properties. Especially good 20 antihypertensive properties have been found in compounds of formula (II) where n is 9 and X is R1, R10 -C - C-. According to the feature of the invention, the compounds of formula (I) wherein R_4 is OH and the compounds of formula (II) are made by a Grignard reaction, in which an imidazolythatone of the form wherein $R_1,R_2,$ and R_3 are as defined before, is reacted with an anylality! magnesium halide derivative or any 35 35 magnesium halide derivative of the formula: wherein $R_{\rm b}$, $R_{\rm s}$ and $R_{\rm r}$ are as defined before, n' is 0 to 5 and Hal is a halogen atom to give compounds of the formula (III) wherein R_1 , R_2 , R_3 , R_6 , R_6 , R_7 and n' are as before. The arylalkylmagnesium halide derivative can be, for example, an arylalkylmagnesiumbromide derivative, which is prepared by reacting the corresponding arylalkylbromide derivative with magnesium. Suitable solvents for the reaction include a variety of ethers, preferably tetrahydrofuran. The enviality imagnesium has 55 lide derivative is prepared in the usual way by adding the arylalkylmagnesiumhalide derivative in a suitable solvent, e.g. tetrahydrofuran, dropwise onto magnesium turnings covered by tetrahydrofuran, at the boiling point of the reaction mixture. When the magnesium turnings have reacted, the mixture is cooled slightly and

the 4-imidezole derivative is added in solid form in small portions or in tetrahydrofurane solution. After the addition, the reaction mixture is refluxed until all of the 4-imidezole derivative has reacted. The reaction time

60 varies between one and five hours.

GB 2 101 114 A

Another process for the preparation of compounds of formula (III) is a Grignard reaction in which a compound of the formula (IV)

10 wherein $R_1 \cdot R_2$ and n' are as before, is reacted with a compound of the formula

RyMgHal

20

35

15 wherein R_3 is an alightor anyliae defined before and Hal is halogen. Yet another process for the preparation of compounds of formula (III) is a Grignard reaction in which an imidazole carboxytic acid alityl easer, preferably the methyl eater of the formula

3

13

20

36

45

erain R_1 and R_2 are as before, is reacted in a first step with a Grignard reagent of the formula

wherein R_b , R_b , R_t and n' are as before, to give a compound of formula (IV), which in a second step wethous isolation is reacted with a Grignard reagent of the formula

R₃MgHal

wherein R₃ is as defined before.

Compounds of formula (I) wherein R_4 is H can be prepared by reduction of compounds of formula (EI)

40 wherein n' is 0 with hydrogen. A suitable catalyst is e.g. palladium-on-carbon.

Unsaturated compounds of formula (I) wherein R₃ is = CH₂, = CH - CH₃,

wherein R_1 , R_2 , R_6 , R_6 , R_7 are as defined before, R_3 is an alkyl or anyl as defined before and n' is 0 to 5, to give 55 a compound of the formula (V)

wherein R is an alkyl; or wherein R_{12} and R_{14} can be combined to form a keto group, or R_{13} and R_{16} can be combined to form a keto group, or both R_{12} and R_{14} and R_{13} and R_{16} can simultaneously form keto groups, or

55

0

11

-O-C-R.

55

5

30

40

55

wherein R₁, R₃, R₃, R₄, R₅, X and n are defined as before. Reagents capable of converting the deplicable starting material to the corresponding imidazole include NH₃ + CH₂O for a source of ammonia and formaldehyde):

25 the boiling point of formamide for a period of time ranging from one to five hours.
Yet another process for the preparation of the compounds of formula (I) and (II) comprises resemble formamide with a benzene derivative of the formula:

23

B

20

73

60

wherein R_b , R_b , R_r , R_e and n are as defined hereinabove, and Q is a radical of formula .

wherein R is a substituted and unsubstituted alkyl, arylalkyl or aryl group, and R₂. Hal and X are as defined hereinsbove. Preferably the reaction is performed by vigorously boiling the benzene derivative in formamide, the reaction time varying with the particular material employed.

45 Reaction times typically are from 30 minutes to 8 hours. Obviously, the formamide treatment will be followed by reaction with an appropriate acid (e.g. HCI) when Q in the starting material is

in order to obtain the corresponding compound of formula (I) and (II).

60 is employed, then the formamide treatment will be followed by hydrogenation, thus affording the desired compound of formula (I) and (II).

7	GB 7	WI 114 A
À further proce corresponding N	ess for the preparation of the compounds of the formula (I) and (II) comprises by N-acetylated compound of the formula (I) and (II)	drefysing ö
5		
10		
atoms an anyl ra 15 Preferably, th in an aqueous e Yet another p	erylatikytresidue determined by the formula (I) and (II), it is an alkyl group of 1 to 2 adical of 6 to 10 carbon atoms. We hydrolysis is carried out by boiling the starting material, an N-acylated imidaes solution of an inorganic acid until the reaction is completed. Process for the preparation of the compounds of formula (I) and (II) comprises by rial of the formula	ple dermative.
20	En ₂	
25		
30 conveniently or stirring or using carbon and Rar	s defined before and R' is an aryl or alkyl and R'' is an aryl group. The hydrogenest onducted in the presence of a suitable catalyst and under a hydrogen atmospherig metallic sodium in liquid ammonia. Suitable catalysts include platmum oxide, ney nickel. Reaction temperatures vary with the particular starting material emplicatures being 25-70°C.	palladium-or
Yet another r	method for the preparation of the compounds of formula (I) or (II) wherein A is	
-CH-Cl	H- cting a N-trialkylsilylimidazole of the formula	
	<.□	
45	v - \$1-4	
wherein Y is an	n alkyl group, preferably methyl, with an arylalkylhalogenide of the formulae	
	Ry CH-Hal or Ry CH-CH-Hal	
example titani methylene chi	i_0 , R_0 , R_7 , R_0 and R_0 are as before and Hal is a halogen atom, in the presence of a L slum tetrachloride, aluminium chloride or zinc chloride. As solvent can be used folloride or chloroform. The reaction is preferably carried out at room temperature trisls for 6-12 hours.	

GB 2 101 114 A The intermediates of formula (VI) and (VII) can be prepared for example as follows An aldehyde of the formula wherein $R_{f b}, R_{f b}, R_{f r}$ and ${f n}$ are as before, is reacted in alkaline or acidic condit 10 ecetone, to give a compound of the formula (VIII) via direct addol condensation 1, - (10,1),-come = 0-10, (MII) 15 vherein R_e is an alltyl as defined before. which compound in a second step is catalytically reduced to give the corresponding saturated compound of the formule: 20 المناسبة الم (XI) which compound in a third step is regionalectively brominated in methanol to give compounds of features as Another method for the preparation of the compounds of the general formula (VII) is the regionalisticave alkylation process of ketones in which for example a halide compound of the formula (X) 6 (CI), 1 - Lu-m1 is reacted with a trimethylsilylenolether derivative of the general formula (XI) **OTMS** OCU Re-CH = C-CH3 40 wherein R_0 is an alkyl as before, in the presence of a Lewis acid, for example zinc (II) chloride, to give a

compound of the formula (XII)

45

60

65

23

25

20

-3

30

40

45

55

is reacted with lithlated N,N-dimethylhydrazone of acetone followed by hydrolysis to give a compound of the general formula (XIV)

(CH₂)_n-(H 1ω1

" - (CH,) " - CH-CH-E CH,

The compound of formula (XII) is further brominated as before to give compounds of the formula (VII). When $R_{\rm B}$ and $R_{\rm J}$ are hydrogen yet another method for the preparation of compounds of the formula (VII)

can be applied. In this method a halide of the general formula (XIII)

$$\frac{H_5}{R_1} = \frac{1}{2} \left(\frac{1}{(CH_2)_{11} - CH_2 - CH_3} \right)$$
(AIV)

(uiv)

_	g apple a c
•	which compounds are brominated as before to give compounds of the formula (VII). According to another method for the preparation of compounds of the formula (VII), compounds of the ormula (VIII) are selectively brominated using as brominating agent for example 2-carboxyethyltriphenylphosphonium perbromids, which has the formula (XV)
5	•
	(C'H')?-b@-CH'CH'C-OH Bu)
10	Yet another method for the preparation of compounds of the formula (VII) is possible via a directed additionable for which for example a compound of the formula (XVI)
15	(av1)
•	is reacted with the compound (XI) in the presence of a Lewis acid विशेष्टकार्थ by deliquistics. to द्वारण a compound of the formula (XVII)
20	
	which compound is further brominated as before to give a compound of the formula (VII) When R ₁₁ is hydrogen, compounds of the formula (VII) can be prepared from compounds of the formula (XVI), wherein these are reacted with 1-lithiated N,N-dimethylhydrazone of methylalkylketone of the form (XVIII)
30	y-α
35	Here in the first step compounds of the formula (XIX) are achieved.
40	$\mathbf{a}_{p} = (\mathbf{c}_{\mathbf{a}^{1}})^{n} - \mathbf{c}_{\mathbf{a}^{1}} = \mathbf{c}_{\mathbf{a}^{1}} \cdot (\mathbf{c}_{\mathbf{a}^{2}} \mathbf{a}^{2}) $ $(\mathbf{x}_{1} \mathbf{x}_{2})$
	the bromination of which compounds are performed following the method above. The preparation of compounds of the general formula (VII) can be accomplished from compounds of its general formula (XVII) by hydrogenation of the carbon-carbon bouble bond as well. The bromination in the second step leads to compounds of the formula (VII). Alkylation of compounds of the general formula (XVII) when R ₂ and R ₁₀ are hydrogen can be accomplished, too. In this method a compound of the formula (XX)
	$\frac{R_{5}}{R_{5}} = \frac{1}{CH_{2}} \frac{1}{n} - CH - \frac{1}{C} - CH_{3} \qquad (acc)$
50	
	is reacted with an alkylation reagent such as dialkyllithiocuprate (XXI) which undergoes 1,4-conjugate addition

GB 2 101 114 A provides further enother method for the preparation of compounds according this in condensation is performed for example in equeous eloohot catalysed by sodium hydroxide. A or their vinylogues have the general formulae (XXIII) and (XXIV) 5 10 (XXIV) (XXIII) es unsaturated betones of the formulae (XXVI) and (XXVII) In the first step this condensation gi 15 20 (XXVI) (XXX) which compounds are then hydrogenated to the end products according to the formulae (XXVIII) and (XXVIII) (XXVIII) (XXVII) 35 As stated herein above, the compounds of the general formula (I) and (II) and their non-toxic. pharmaceutically acceptable acid addition salts have valuable pharmacological properties and have been 40 found to possess excellent antihypertensive properties. Tests have shown that they also possess other pharmacological properties as well, for example, antithrombotic activity. Furthermore, antimycotic and antifungal properties have been found, too. The processes described above for the preparation of compounds of formula (II) wherein X is R11 R10 45 result mainly in the trans isomer of the compound. The trans isomer can be converted to the cis isomer 50 according to known methods, e.g. by heating it in the presence of an acid or by irradiating it with ultraviolet light. Administration of isomeric compounds of formula (i) and (ii), their non-toxic, pharmacoutically acceptable acid salts or mixtures thereof may be achieved parenterally, intravenously or orally. Typically, an effective amount of the derivative is combined with a suitable pharmaceutical carrier. As used herein, the term 55 "effective amount" encompasses those amounts which yield the desired activity without causing adverse side-effects. The precise amount employed in a particular situation is dependent upon numerous factors such as method of administration, type of mammal, condition for which the derivative is administered, etc., and of course the structure of the derivative. The pharmaceutical carriers which are typically employed with the derivatives of the present invention

60 may be solid or liquid and are generally selected with the planned manner of administration in mind. Thus, for example, solid carriers include lactose, sucrose, gelatin and agar, while liquid carriers include water, syrup, peanut oil and olive oil. Other suitable carriers are woil-known to those skilled in the art of

GB 2 101 114 A determined by the following procedure. Sprague-Dawley rats of normal weight were first enesthesized with urethene. After this, the femoral entery was connected by way of a polyethylene tube with a blood pres transducer. The test substance was then injected into the femoral vein and the blood pressure and the pulse frequency were registered with a recorder. In a further test for anti-hypertensive properties unanesthetized Water spontaneous hypertensive ra (SHR) were used. The test derivative was administered percently by way of a tube into the stemach. The blood pressure was measured from the tail using an indirect bloodless method In another experiment 3 months old spontaneous hypertensive male rats were used to test the anti-hypertensive properties during a period of 4 weeks. The test derivative was adminis red daily to each 10 rat in the drinking water and the blood pressure of the tall was measured by a standard electric m The antithrombotic activity was investigated in vitro. The inhibiting activity of the compounds ago ADP- and collagen-induced aggregation of thrombocytes was measured. In the test thrombocytes from a cow was used. To 1.2 ml of plasme containing 250000 thrombocytes/mm³ were added 50ul of a solution of the compound to be tested. After 10 min incubation either ADP or collegen was added. The aggregate 15 the thrombocytes was turbidimetrically determined at $\lambda = 805$ n m. 35 The antimicrobial activity was determined in witro according to a qualitative test for analogous antifungal activity, using the ager diffusion method, against the following standard organisms. Standard organisms. us aureus, Streptococcus pyogenes, Escherichia coli, Proseus mirebilis, Paeudomones eeruginosus, Candida albicans and Aspergillus niger. The antifungal activity was determined in vitro against the following fungi: Trichophysen rubrum Trichophyton mentagrophytis, Microsporum canis, Epidermophyton floccotum, Chrysosporum, Ca albicans, Candida guilliermond; and Saccaromyces cerevisiae. The fungi were cultured by planing on an ag nutrient medium. The compound to be tested was added before the incubation. A measure of the efficiency of the compound tested is the radius of the circle, within which the growth of the fungs has been inhibite Acute toxicity was determined by using female mice of NMRI-Strain with an age of about 7 months and weighing 30-40 g. The administration of the test compound was i.v. Thus, the compound 4-[2-12,6-dimethylphenyl)-1-methylethenyl)midazole, which has a LD_{to} value great than 30 mg kg i.v., was found in the blood pressure study with anesthetized rats of normal weight described above to cause a registrable lowering of the blood pressure at a dose of 3 μg kg i.v. At a dose of 10 μg kg i.v 30 the blood pressure lowering was quite clear and at a dose of 100-300 μg/kg i.v. the reduction of the blood pressure was on an average 38 %. The duration of the effect was at least 30 minutes (after which time the The compound 4-[2-(2,6-dimethylphenyl)-1-methylethyl]imidazole caused a blood pressure lowering of 20 determination was interrupted). per cent measured 30 minutes after the administration at a dose of 100 µg kg. The compound 4-((a-methyl)-(2,6-dimethylbenzyl)imidazole caused a blood pressure lowering of 50 % at a The compound 4-[(a-methyl)-2,3-dimethylbenzyl) imidazole caused a blood pressure drop of 55% at 10 dose of 30-100 µg kg. In the Examples below, where 'H-NMR spectrum shifts are presented, the NMR spectra were determined ma/ka. 40 with a Perkin-Elmer R 24 or a Bruker WP80DS apparatus using an external tetramethylsilane standard, from which the presented chemical shifts (h, ppm) are tabulated. The letters s, d, t and m are used to indicate a singlet, doublet, triplet or multiplet, respectively and coupling constants in hertz when given. In the same connection, the number of hydrogen atoms is also stated. The compounds which are indicated as bases are tested in deuterium methanol, deuterium acetone or deuterium chloroform, while the values for compounds 45 which are indicated as hydrochlorides were determined in deuterium oxide. The presented 13C-NMRspectrum were determined with a Bruker WP80DS apparatus. The mass-spectra were determined with a Perkin-Elmer RMU-6E apparatus using direct inlet system. The temperature employed was the lowest temperature needed for the evaporation of the compound as base. In the examples the strongest and the most essential fragment-ions from a structural viewpoint are given as 50 m/e values. In parenthesis is given the intensity of the fragment-ion in relation to the main peak. 50 4-(ւլ.ս-bis(2-methylphenyl)hydroxymethyl]-5-methylimidazole 4.9 g (0.2 mol) of dry magnesium turnings are covered with 60 ml of dry tetrahydrofuran. The mixture is 55 heated to boiling and a solution of 34 g (0.2 mol) of 2-bromotoluene in 50 ml dry tetrahydrofuran is added dropwise at such a rate that a smooth reaction is maintained. After the addition is complete, the reaction mixture is refluxed for about 30 minutes until the magnesium turnings no longer react. The reaction mixture is then cooled to about 50°C and 9.3 g of 5-methyl-4-imidazole carboxylic acid methyl ester are added in small portions. After the addition is complete, the mixture is refluxed for another 2 hours and the solvent is 60 then distilled off to give about half of the original volume. The mixture is cooled and poured into 300 ml of cold water containing 15 ml of concentrated sulfuric acid, with agitation. The stirring is continued for an

additional 16 minutes and the mixture is then filtered. The precipitate, filtered from the acidic water, which is

. handa materia nethanal natution. After

••	GB 2 101 114 A	12
15		
•	MS: 292 (55 %), 274 (69 %), 269 (100 %), 232 (7 %), 217 (9 %), 201 (62 %), 199 (77 %), 167 (18 %), 139 453 %)	
	EXAMPLE 2	
	#/n.n-bis/2-methylpheny//hydroxymethylphinidarole A Grignard reagent is prepared from 68.4 g of o-bromotolusne and 8.6 g of Mg-turnings in 260 ml of The To this solution 12.6 g of 4-imidarole carboxylic acid methylester are added at 50°C and the reaction misture	\$
•	is refluxed for 5 hours. The mixture is then poured into cold water, which includes 60 ml of conc. HCI, The hydrochloride of the	10
?0	product is filtered off, washed with charlocal mail and a water-athenol with sodium/hydroxide, m.p. 138 140°C 178-179°C. Liberation of hydrochloride is achieved in water-athenol with sodium/hydroxide, m.p. 138 140°C "H-NMR (HCI-salt): 1.9 (s, 6H), 4.6 (s, 3H), 6.7 (s, 1H), 7.0 (s, 8H), 8.7 (s, 1H)	
	EXAMPLE 3	
15	4-fix a-dipheny()hydroxymethyl-5-methylimidatole	:5
20	MS: 264 (80 %), 246 (78 %), 231 (28 %), 218 (20 %), 204 (9 %), 167 (100 %), 103 (01 %), 103 (01 %)	100
	EXAMPLE 4 4-[(n-(2-methylphenyl))-2-methylbensyl)imidarole	
25	The starting material, 4-(u, n-bisl2-methytphenylinydroxymennyl-brusarus is discussed as the starting material, 4-(u, n-bisl2-methytphenylinydroxymennyl-brusarus) in a hydrogen atmosphere at acid. 100 mg of Pd C are added and the reaction mixture is stirred vigorously in a hydrogen atmosphere at acid. 100 mg of Pd C are added and the reaction is completed. The mixture is then filtered and distributed to a smaller volume. 70 about 60°C until the reaction is completed. The mixture with 20 ml portions of chloroform. The aqueous	75
	phase is made attaine with NaCH and extracted with countries. The solution is graporated to dryness	
30	extracts are washed with water (1 × 10 mi) and disease from ethyl acetate-isopropancial im.p. 245-254 C Yield 93 %, m.p. 228-231 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.p. 245-254 C Yield 93 %, m.p. 228-231 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.p. 245-254 C Yield 93 %, m.p. 228-231 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.p. 245-254 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.p. 245-254 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.p. 245-254 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.p. 245-254 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.p. 245-254 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.p. 245-254 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.p. 245-254 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.p. 245-254 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.p. 245-254 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.p. 245-254 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.p. 245-254 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.p. 245-254 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.g. 245-254 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.g. 245-254 C (from ethanol). Hydrochloride from ethyl acetate-isopropancial im.g. 245-254 C (from ethanol).	30
	EXAMPLE 5	
	4-(u-phenylbenzyl)-5-methylimidazole	
35	The compound is prepared by reduction of 4-(دردا-diphenyl)hydroxymethyl-5 methylimidaæise هواهدينه ه palladium-on-carbon catalyst as described in Example 4.	æ
	EXAMPLE 6	
40	4-(u-phenylbenzyl)-5-methylimidazole The compound is prepared from 4-(u, u-diphenyl)hydroxymethyl-5-methylimidazole according to the method in Example 4. Yield 69 %, m.p. 198-204°C (from ethanol).	40
	¹H-NMR: 1.6 (s, 3H), 4.5 (s, 1H), 5.3 (s, 1H) 6.8 (s, 10H), 7.3 (s, 1H)	
	EXAMPLE 7	45
4	5 4-(u-(2-methylphenyl)-u-(2-methylbenzyl))-5-methylimidazole The compound is prepared according to the method in Example 4 using 4-(u,u-tus-(2-methylphenyl)) to the method in Example 4	
	methylphenyl)hydroxymethyl]-5-methylimidazole as starting material. Yold 70	
	water-ethanol). ¹ H-NMR: 1.4 (s, 3H), 1.8 (s, 6H), 4.6 (s, 1H), 5.35 (s, 1H), 7.1 (m, 8H), 7.16 (s, 1H)	50
5	0	30
	EXAMPLE 8 4-[[(2,3-dimethy/pheny/]methy/]hydroxymethy/]imidazole, 1-(4-imidazoly/)-1-(2,3-	
	For the preparation of 2.3-dimethylmagnesium promide in the mat step. 4.5 g of any	55
5	is are covered with 50 ml of dry tetrahydrufuran. The mixture is heated to boiling and a solution of 37 g of 2.3-dimethylbromobenzene in 50 ml dry of	
		ı
_	In the same way in another flask of matnylmagnesium formide is prepared.	60
•	turnings and 9.5 g of methylpromide in terranyarollars. Yet another flask of 12.6 g of 4-imidezolecarboxylic acid methylester, is added to 100 ml of dry Yet another flask of 12.6 g of 4-imidezolecarboxylic acid methylester, is added to 100 ml of dry	

GB 2 101 114 A 13 water containing 50 ml of concentrated sulfuric acid, with agitation. The stirring is continued for an additional 15 minutes and the mixture is then filtered. The pH of the filtrate is adjusted slightly basic and the mixture is extracted three times with \$9 ml particular of methylene chloride. The combined methylene chloride extracts are washed with water and evaporated to 5 dryness. The residue which contains crude 4-([a-(2,3-dimethylphenyl)-a-methyl]hr/droxymethyl] imidesate of further purified column chrometographically in allicagel using chloroform methenol as eluent. 1-44imidazolyl)-1-(2,3-dimethylphenyl)-ethylene is then obtained from the above product by heating it will potassiumhydrogen aulphute at 136°C. H-NIMR (HCF-calt): 2.104 (s, 3H), 2.313 (s, 3H), 5.187 (s, 2H), 5.358 (s, 1H), 6.106 (s, 1H), 7.03-7.22 (m, 4H), 6.56 (s, 1H) ¹³C-NMR (HCI-saft); Signats et ppm: 18.073, 21.857, 118.789, 119.456, 127.961, 129.475, 132.238, 135.802. 136.498, 136.892, 137.921, 139.949, 140.070 Melting point as base: 137-140°C 4-(a-methyl-2,3-diemthylbenzyl)imidezole is obtained via hydrogenation with palladium-on-carbon catalyst 15 in 2-N HCl according to the method described before. H-NMR (HCl-salt): 1.708 (d, 3H), 2.370 (broad a, 3H), 4.668 (q, 1H), 4.833 (s, 2H), 7.079-7.263 (m, 3H), 7.351 (s, 1H), 8,780 (s. 1H) ¹³C-NMR (HCI-salt): Signals at ppm: 16.529, 21.917, 22.462, 34.662, 117 881, 126.660, 128.365, 131.679, 135.650, 136.952, 140.161, 140.163, 142.855 By the same method for example the following compounds were prepared: 4-(a-methyl-2,6-dimethylbenzyl)imidazole 4-(a-ethyl-2,3-dimethylbenzyl)imidazole 4-(a-butyl-2-methylbenzyl)imidazole 4-(a-methyl-2,3-dimethylbenzyl)-2-methylimidazole **EXAMPLE 9** 4-(2,6-dimethylphenyl)-3-buten-2-one 13.4 g (0.1 mol) of 2,6-dimethylbenzaldehyde, 100 ml of acetone, 100 ml of water and 2 g of calcium hydroxide are mixed together and refluxed for about 20-25 h with agitation. The precipitate is filtered off 30 from the cold reaction mixture. 1 i of ice water is added to the filtrate with agitation. The product is 30 crystallized at a yield of about 90 %. M.p. of the recrystallized product: 34-35°C. ¹H-NMR: 7.55 (1Hd, 16.5), 7.00 (3Hs), 6.26 (1Hd, 16.5), 2.37 (3Hs), 2.31 (6Hs) **EXAMPLE 10** 35 4-(2,6-dimethylphenyl)-2-pentanone To a mixture containing 20 g of Cul and 50 ml of tetrahydrofuran (THF) are added 105 ml of methyllithium dropwise during with agitation in a nitrogen atmosphere at a temperature of 0°C or lower until the yellow precipitate barely dissolves. Then 8.7 g of 4-(2,6-dimethylphenyl)-3-buten-2-one in 50 ml of THF are added slowly at 0°C. The stirring is continued for an additional 2 h with a gradual increase of the temperature to + 40 25°C. The reaction mixture obtained is hydrolysed with 300 ml of a solution of NH_aCl. The ether is removed, dried and evaporated to give the crude product. ¹H-NMR: 6.85 (3Hs), 3.78 (1Hq + t, 7.5), 2.76 (2Hd, 7.5), 2.34 (6Hs), 1.99 (3Hs), 1.27 (3Hd, 7.5) According to the same method, the compound 4-phenyl-2-pentanone was prepared. ¹H-NMR: 7.10 (5Hs), 3.26 (1Hq+t, 7.5), 2.62 (2Hd, fine structur), 1.94 (3Hs), 1.20 (3Hd, 7) 45 Similarly 4-(3,4-dimethylenedioxyphenyl)-2-pentanone 'H-NMR: 6.62 (3H, s), 5.83 (2H, s), 3.20 (1Hq + t, 7), 2.67 (2H, d7), 2.04 (3Hs), 1.26 (3Hd7) **EXAMPLE 11** 1-bromo-4-(2.6-dimethylphenyl)-2-pentenone To 3.8 g of 4-(2,8-dimethylphenyl)-2-pentanone in 25 ml of dry methanol 1.04 ml of bromine are added 50 dropwise rapidly at a temperature not higher than + 5°C, Stirring is continued until the bromine colour disappears, while the temperature slowly rises to + 20°C. After evaporation the product is obtained at a vield of at least 70 % H-NMR: 6.98 (3Hs), 3.80 (1Hm), 3.67 (2Hs), 3.02 (2Hd), 2.35 (6Hs), 1.33 (3Hd, 7) According to the same method the compounds 1 bromo-4-phenyl-2-pentanone and 1-bromo-4-(2,6-55 dimethylphenyl)-3-methyl-2-butanone were prepared. Similarly using two equivalents of bromine: 1-bromo-4-(2-bromo-4,5-methylenedioxyphenyl)-2-pentanone ¹H-NMR: 6.9 (1H, s), 6.67 (1H, s), 5.87 (2Hs), 3.80 (2Hs), 2.9 (3Hm), 1.19 (3Hd7) **EXAMPLE 12** 4-12-12.6-dimethylphenyllpropyllimidezole

14	√GB 2 101.114 A	14
`	This is dissolved in ethyl ecetate and HC/ethylacetate is added. The product is evaporated to dryness. Washed with other, dissolved in water, neutralized with NeHCO ₂ and extracted with methylane change. The Proporation residue is dissolved in ethyl acetate and the final product is precipitated as existed or	
5	hydrochloride. M.p. of the hydrochloride 194-196°C. 'H-NHMR (HCI-ealt): 8.70 (1He), 6.9 (4He), 3.65 (1Hm), 3.21 (2H, d 8) 2.39 (6H broad e), 1.46 (3Hd 7) According to the same method, the following compounds were prepared: 4-[2-(2,6-dimethylphenyl)-1-methylethyl]imidazole. M.p. of the oxistee 161-6°C. 'H-NMR (oxiste): 9.75 (1H broad s), 7.05 (1He), 7.00 (3He), 3.0 (3Hm), 2.20 (6He), 1.31 (3Hd)	\$
10	4-(2-phenylpropyl)imidazole (se oxalate) 'H-NMR: 8.52 (1Hs), 7.22 (5Hs), 6.97 (1Hs), 3.05 (3Hm), 1.35 (3Hd) M.p of the oxalate: 166-166°C.	10
	4-[2-(3,4-methylenedickyphenyliphopyphilosolo 'H-NMR (as oxalate): 8.70 (1H, s), 7.11 (1H, s), 6.78 (3H, m), 5.95 (2H, s), 3.08 (3H, m), 1.40 (3H, d) M.p. of oxalate: 154-156°C. 4-[2-(2,6-dimethylphenyl)butyl}imidazolo	85
13	M.p. of oxiste 176-9°C. 14-NMR: 8.25 (1H, broad s), 6.95 (3H, s), 6.68 (1H, s), 3.2 (3H, m), 2.45 (3H, s), 2.6-2 0 (2H, m), 2.06 (3H, s), 1.87 (3H, s)	
20	4-[2-(2-bromo-4,5-methylenedioxyphenyl)propyl)imidazole EXAMPLE 13 4-(2,6-dimethylphenyl)-3-methyl-3-buten-2-one	70
25	4-(2,6-dimethylphenyl)-3-methyl-3-buten-2-one A mixture of 13.4 g of 2,6-dimethylbenzaldehyde and 15 ml of 2-butanone is saturated with gaseous HCI with stirring. The starting temperature is 0°C, and it is raised to 20-25°C in 2 h. The reaction mixture is poured into 0.5 l of cold water, extracted with toluene and washed with a NaHCO ₂ -solution. The dried toluene extract is filtered, toluene and free 2-butanone are distilled off. The product is obtained by crystallization form di-isopropylether. M.p. 43-44°C. 1H-NMR: 7.38 (1Hs), 6.98 (3Hs), 2.42 (3Hs), 2.11 (6Hs), 1.59 (3Hd, 1.4)	25
		30
	EXAMPLE 14 1-bromo-4-(2,6-dimethylphenyl)-3-methyl-3-buten-2-one To a mixture of 3.8 g of 4-(2,6-dimethylphenyl)-3-methyl-3-buten-2-one in 50 ml of THF a solution of 13 g of 2-carboxyethyltriphenylphosphoniumperbromide in 50 ml of THF is added dropwise at room temperature Stirring is continued for another 2 h. 200 ml of water and 100 ml of ligroin are added. The organic layer is washed with a Na ₂ CO ₃ -solution and water. After filtration and 5-5 g of crude product containing washed with a Na ₂ CO ₃ -solution and water. After filtration and 2-one are obtained.	n
33	washed with a rea, CO3-sciulidir and washed washed with a rea, CO3-sciulidir and wash	
40	EXAMPLE 15 4-[2-(2,6-dimethylphenyl)-1-methylethenyl]imidazole E-isomer The compound is prepared according to the method described in Example 12, except that 1-bromo-4-(2,6-dimethylphenyl)-3-methyl-3-ben-2-one is used instead of 1-bromo-4-(2,6-dimethylphenyl)-2-pentanone.	40
49	M.p. of the hydrochloride 260-262°C. H-NMR: 8.82 (1H, d), 7.36 (1H, d), 7.20 (1H, broad s), 7.08 (3Hs), 2.20 (6Hs), 1.82 (3H,d,1.2)	45
	EXAMPLE 16 4-((a-methyl)-2,6-dimethylbenzyl/imidazole To a mixture of N-(trimethylsilyl)imidazole (1,4 g) and titanium tetrachloride (1.6 ml) in dry chloroform (20 ml) a solution of 1-chloro-1-(2,6-dimethylphenyl) and titanium tetrachloride (1.6 ml) in dry chloroform (10 ml) was added. After ml) a solution of 1-chloro-1-(2,6-dimethylphenyl) and titanium tetrachloride (1.6 ml) in dry chloroform (10 ml) was added. After ml) a solution of 1-chloro-1-(2,6-dimethylphenyl) and titanium tetrachloride (1.6 ml) in dry chloroform (20 ml) and titanium tetrachloride (1.6 ml) in dry chloroform (20 ml) and titanium tetrachloride (1.6 ml) in dry chloroform (20 ml) and titanium tetrachloride (1.6 ml) in dry chloroform (20 ml) and titanium tetrachloride (1.6 ml) in dry chloroform (20 ml) and titanium tetrachloride (1.6 ml) in dry chloroform (20 ml) and titanium tetrachloride (1.6 ml) in dry chloroform (20 ml) and titanium tetrachloride (1.6 ml) in dry chloroform (20 ml) and titanium tetrachloride (1.6 ml) in dry chloroform (10 ml) was added. After ml) a solution of 1-chloro-1-(2,6-dimethylphenyl) and titanium tetrachloride (1.6 ml) in dry chloroform (10 ml) was added. After ml) a solution of 1-chloro-1-(2,6-dimethylphenyl) and titanium tetrachloride (1.6 ml) in dry chloroform (10 ml) was added.	50
5	ml) a solution of 1-chloro-1-(2,8-dimethylpheny	
5	M.p. of the hydrochloride: 208-10°C. M.p. of the hydrochloride: 6 CD ₃ OD 139.8 (1C, s), 139.1 (1C, s), 137.6 (2C, s), 134.9 (1C, d), 130.8 (2C, d). 12C-NMR (as hydrochloride): 6 CD ₃ OD 139.8 (1C, s), 139.1 (1C, s), 137.6 (2C, s), 134.9 (1C, d), 130.8 (2C, d).	56

00 2 101 114 A 15 CLAIMS 1. Substituted imidazoles of the general formula 10 wherein R_1 is H, an alkyl of 1 to 4 carbon atoms or - CH₂OH; R_2 is H or CH $_2$: R_3 is - CH $_3$ - CH $_3$ CH $_3$ 20 -CH2CH2CH3 _CH₂ -CH-CH₂ -CH₂CH₂CH₂CH₃ -CH₂CH-CH₃, or and R4 is H or OH; or R3 and R4 together represent =CH₂, =CH-CH₃ =CH-CH₂-CH₃ =C-CH₃ or =CH-CH₂CH₂CH₃: 30 R11 R10 R. R. X is -CH-CH or $-C_- = C-$; R_b , R_a and R_2 , which can be the same or different, are H, $-CH_3$. $-CH_2CH_3$. 35 halogen, OH or -OCH₃ or R₆ is hydrogen and R₆ and R₇ together form an -O -CH₂-O-bridge between adjacent carbon atoms in the phenyl group; -CHR₆- is -CH₂-, -CHCH₃-, -CHCH₃-, -CHCH₃-, -CHCH₃-, CH₃

-CH(-CH₂CH₃CH₃)-, -CH(CH(-CH-CH₃-, -CH(-CH₂CH₃CH₃CH₃)-, or 40 $C=CH_2$, $C=CH-CH_3$, $C=CH-CH_2CH_3$, $C=C-CH_3$, or C=CH-CH2CH2CH3; Rais H, -CH3, -CH2CH3, -CH2CH2CH3. CH₃
-CH CH₃, -CH₂CH₂CH₂CH₃ or OH; R₁₀ is H, -CH₃, -CH₂CH₃, 50` CH₃
55 - CH₂CH₂CH₃, - CH CH₃, or - CH₂CH₂CH₂CH₃; R₁₁ is H, - CH₃. CH₃
-CH₂CH₃, --CH₂CH₂Ch₃, --CH-CH₃ or -- CH₂CH₂CH₂CH₃; n is 0 to 4; provided that

- when R4 is OH, R1 is H or CH3 and R3 is

GB 2 101 '14 A - when R_1 , R_2 and R_4 all are hydrogen and R_2 is 5 then R_b, R_b, R₇ are not all simultaneously hydrogen. - Re and Re are not simultaneously hydrogen, and -R₁₁ and R₁₉ are not simultaneously hydrogen; and their non-toxic pharmaceutically acceptable acid addition salts. 2. A compound of formula (I) or (II) according to Claim 1 wherein each of R_b, R_b and R_b, which can be the same or different, is hydrogen, methyl, ethyl or halogen. 3. A compound of formula (I) according to Claim 1 or 2 wherein R_e is hydrogen and R_e is maxing, extra 1. propyl, isopropyl or butyl. A compound of formule (I) according to Claim 3 wherein R, is methyl. 5. A compound of formula (II) according to Claim 1 or 2 wherein X is 15 R,, R, -C = C - and R₁₀ is hydrogen. TO. 20 6. A compound of formula (II) according to Claim 1 or 2 wherein X is R,,R,0 75 25 -C=C- and R₁₁ is hydrogen. 7. A compound of formula (II) according to Claim 1 or 2 wherein X is 30 30 Ra Ra -CH-CH- and R₀ is hydrogen. 8. A compound of formula (II) according to Claim 1 or 2 wherein X is 35 35 Re Ro -CH-CH- and Ra is hydrogen. 40 9. A compound of formula (II) according to any of claims 1, 2 and 5-8 wherein R₁ is hydrogen or methyl 10. A compound of formula (II) according to any of claims 1, 2 and 5-9 wherein n is 0 or 1. 11. A compound of formula (II) according to Claim 1 wherein each of R_b, R_b and R_J, which can be the same of different, is hydrogen, methyl, ethyl or halogen; R_1 is hydrogen or methyl; R_2 is hydrogen or methyl, 45 R_{B} or R_{11} is methyl, ethyl or isopropyl, R_{B} or R_{10} is hydrogen, and n is 0. 12. 4-[u,u-bis(2-methylphenyl)hydroxymethyllimidazole and its non-toxic pharmaceutically acceptable acid addition salts. 13. 4-[(a-(2-methylphenyl)]-2-methylbenzyl]imidazole and its non-toxic pharmaceutically acceptable acid addition salts. 14. 4-(u-phenylbenzyl)-5-methylimidazole and its non-toxic pharmaceutically acceptable acid addition 50 salts. 15. 4-[(α-(2,6-dimethylphenyl)]-α-methyl|hydroxymethyl|imidazole and its non-toxic pharmaceutically acceptable acid addition salts. 16. $4\cdot[(\alpha\cdot(2,3\cdot\dim\operatorname{ethylphenyl})]\cdot\alpha\cdot\operatorname{methyl]}$ by droxymethyl] imidazole and its non-toxic pharmaceutically 55 55 acceptable scid addition salts. 17. 4-[cs,cs-bis(2-methylphenyl)hydroxymethyl]-6-methylimidarole and its non-toxic phermacoutically acceptable acid addition salts. 18. $4-[(\alpha-(2-methylphenyl)-2-methylbenzyl-5-methylimidazole and its non-toxic pharmaceutically$ acceptable sold addition salts. 60 19. 4-[(a-methyl)-2,6-dimethylbenzyl]imidazole and its non-toxic pharmaceutically acceptable acid addition salts. 20. 4-[(a-methyl)-2,3-dimethylbenzyl]imidazole and its non-toxic pharmaceutically acceptable acid

GB 2 101 114 A 17 22. 4-[(a-butyt)-2,3-dimethylbenzyl]imidezole and its non-toxic phermaceutically acceptable sold add-23. 4-((a-methyl)-2,3-dimethylbenzyl)-2-methylimidezole and its nontoxic pharmaceutically acceptable tion salts edid addition safts 24. 4-{(a-propyl)-2-methylbenzyl}imidazole and its non-toxic phermeceuticsPy acceptable acid addition 4-(a-methyl)-2-methylbenzyl)imidezole and its non-toxic phermaceutically acceptable sold address salts. 25. salts 4-((a-methyl)-2,5-dimethylbenzyl)tmidezole and its non-toxic pharmaceutically acceptable acid 26. 27. 4-(a-sthyl-a-(3-methylphenyl)-hydroxymethyl)imidszole and its non-toxic pharmaceutically 10 addition salts. acceptable acid addition salts. 28. 4-(a-butyl-u-(2,3-dimethylphonyl)-hydroxymethyl)imidazole and its non-toxic pharmaceutically acceptable acid addition salts. 29. 4-(a-methyl-a-(2,3-dimethylphenyl)-hydroxymethyl)-2-methylimidazole and its non-toxic pharmacautically acceptable acid addition salts. 30. 4-(a-propyl-a-(2-methylphenyl)-hydroxymethyl)imidezole and its non-toxic pharmaceut-cally acceptable acid addition salts. 31. 4-(a-methyl-a-(2-methylphenyl)-hydroxymethyl)imidazole and its non-toxic phermaceutically 20 20 acceptable soid addition salts. 32. 4-(a-methyl-a-(2,5-dimethylphenyl)-hydroxymethyl)imidazole and its non-toxic pharmaceutically acceptable acid addition salts. 33. 4-[2-(2,6-dimethylphenyl)-1-methylethyl]imidazole and its non-toxic pharmaceutically acceptable 34. 4-[2-(2,6-dimethylphenyl)propyl]imidazole and its non-toxic pharmaceutically acceptable acid addiacid addition salts. 35. 4-[2-(2,6-dimethylphenyl)-1-methylpropyl]imidazole and its non-toxic pharmaceutically acceptable tion salts. 36. 4-[2-(2,6-dimethylphenyl)-2-hydroxyethyl]imidazole and its non-toxic pharmaceutically acceptable acid addition salts. 10 37. 4-(2-phenylpropyl)imidazole and its non-toxic pharmaceutically acceptable acid addition salts. 30 acid addition salts. 38. 4-[2-(2,6-dimethy/phenyl)-1-methy/ethenyl]imidazole and its non-toxic pharmacoutically acceptable acid addition salts. 39. 4-[2-(2,6-dimethylphenyl)-1-propenyl[imidazole and its non-toxic pharmacoutically acceptable acid 35 35 addition salts. 40. 4-(2-methyl-4-phenyl-1-butenyl)imidazole and its non-toxic pharmaceutically acceptable acid addi-41. 4-[2-(4-chlorophenyl)-1-methylpropyl]imidazole and its non-toxic pharmaceutically acceptable acid tion salts. addition salts. 42. 4-[5-(2,6-dimethylphenyl)-1-methyl-1-pentenyl]imidazole and its non-toxic pharmaceutically 40 acceptable acid addition salts. 43. 4-[3-(2,6-dimethylphenyl)-2-methyl-1-propenyl]imidazole and its non-toxic pharmaceutically acceptable acid addition salts. 44. 4-[2-(2,6-dimethylphenyl)-1-ethylethenyl|lmidazole and its non-toxic phermaceutically acceptable 45 45. 4-[α, α-bis-(2,3-dimethylphenyl)hydroxymethyl]imidazole and its non-toxic pharmaceutically 45 acid addition salts. acceptable acid addition salts. 46. 4-[α-(2,3-dimethylphenyl)-2,3-dimethylbenzyl]imidazole and its non-toxic pharmaceutically acceptable acid addition salts. 47. 4-[(a-ethyl)-2,6-dimethylbenzyl]imidazole and its non-toxic pharmaceutically acceptable acid addition 50 60 4-[(α-ethyl)-2,3-dimethylbenzyl]imidazole and its non-toxic pharmaceutically acceptable acid addition salts. 48. 4-[2-(2,3-dimethylphenyl)-1-methylethenyl]imidezole and its non-toxic pharmaceutically acceptable saits. 49. 55 50. 4-[2-(2,6-dimethylphenyl)-1-isopropylethenyl]imidazole and its non-toxic pharmaceutically 55 acid addition salts. acceptable acid addition salts. 51. 4-[2-(2,6-dimethylphenyl)-1-methylethenyl]-2-methylimidezole and its non-toxic pharmaceutically acceptable acid addition salts. 52. 4-[2-(2,6-dimethylphenyl)-1-methylethenyl]-6-methylimidazole and its non-toxic pharmaceutically 60 acceptable acid addition salts. E2 4.(2.(2 &.dichlorophanyl)-1-mathylathanyllimidazole and its non-toxic pharmaceutically acceptable

18 GB 2 101 114 A

25

30

40

50

60

55. 4-(3-(2,6-dimethylphenyl)-1-echyl-1-propenyl (imidarole and its non-toxic phermacinality acceptable acid addition salts.

EAS 9 9 55 MEETS 1 4 MAN 1991 1991

56. 4-(5-(2.6-dimethylphenyl)-1-methyl-1-pentanyl)-5-methyl-midazole and its non-toxic pharmaceutically acceptable acid addition salts.

5 57. 4-(5-(2,6-dimethylphenyl)-1-methylpentyl)imidazole and its non-toxic phermaceuscally acceptable acid addition salts.

58. 4-(4-(2,6-dichlorophenyl)-1-methyl-1-butenyl)imidarole and its non-toxic phermaceutically acceptable acid addition salts.

59. 4-[2-(2,6-dimethylphenyl)-1-ethylethyl)imidazole and its non-toxic phermeceutically acid addition

10 salts.

60. 4-[2-(2,6-dimethylphenyl)-2-ethylethyl)inxidazole and its non-toxic pharmaceutically acceptable acid addition salts.

61. 4-[2-(3,4-methylenedioxyphenyl)propyl)imidazole and its non-toxic phermaceutically acceptable acid addition salts.

*

35

50

55

60

15 62. 1-44-imidazolyl)-1-(2,3-dimethylphenyl)-ethylene and its non-toxic pharmaceutically acceptable acid addition salts.

63. 1-(4-imidazolyl)-1-(2,6-dimethylphenyl)-ethylene and its non-toxic pharmaceutically acceptable acid addition salts.

64. A process for the preparation of a compound of formula (I) as claimed in Claim 1, wherein R_a is QH, 20 which comprises reacting an imidazoly/fixtone of the formula

wherein R_1 and R_2 are as defined in Claim 1 and R_3 is as an alkyl or anyl as defined in claim 1 with an anyl magnesium halide derivative of the formula

35 wherein R_b, R_b R₇ are as defined in claim 1 and Hal is halogen.
65. A process for the preparation of a compound of formula (I) or (II) as claimed in Claim 1, wherein X is

and R_{10} is hydrogen, which comprises reacting a compound of the formula

wherein R_1 - R_2 are as defined in claim 1 and n^2 is 0 to 5 with a compound of the formula

Q8 2 101 114 A 19 68. A process for the preparation of a compound of formula (1) or (II) as claimed in Claim 1, which comprises reacting an imidazote carboxytic acid alltyl eater of the formula erein R $_1$ and R $_2$ are as defined in claim 1 and R' is alkyl in a first step with a Grignard reagent of the formula 15 R_{\bullet} , R_{7} are as defined in claim 1 and n' is 0 to 5, to give a compound of the formula 25 which in a second step is reacted with a Grignard reagent of the formula 30 R₂M₆Hal wherein R₂ is an alkyl or anyl as defined in Claim 1, to give a product, which in the final step is dehydrated to give a compound of formula (I) or (II). 67. A process for the preparation of a compound of formula (I) as claimed in Claim 1 wherein R_a is 35 hydrogen, which comprises catalytic reduction of a compound of the formula 35 40 wherein R_1 , R_2 , R_6 , R_6 and R_7 are as defined in Claim 1 and R_3 is an alkyl or anyl as defined in Claim 1. 68. A process for the preparation of a compound of formula (II) as claimed in Claim 1 wherein X is R11 R10 50 and R_{10} is hydrogen, which comprises dehydration of a compound of the formula 55 80

wherein R_1 , R_2 , R_6 , R_7 are as defined in Claim 1, R_3 is an alkyl or anyl as defined in Claim 1 and n' is 1 to 5.

CH-CH rd $R_{
m e}$ is hydrogen, which comprises catalytic reduction of a compound of the formula of the warried Contraction 10 15 wherein R_1, R_2, R_4, R_5, R_7 and n are as defined in Claim 1 and R_6 is all years in Claim 3 70. A process for the preparation of a compound of formula (II) as claimed in Claim 1 influment II is R., R., -C =C-20 and R_{11} is hydrogen, which comprises reacting as imidazole aldehyde of the formula 25 wherein R_1 and R_2 are as defined in claim 1, with an arallylidenetriphenylphosphoraric of the formula (1000) 34-5 - (1005) - () 35 wherein R_b , R_θ , R_{10} and n are as defined in Claim 1. 71. A process for the preparation of a compound of formula (II) or (III) as claimed in Claim 1, wherein X is 40 Re Re which comprises reacting a N-trialkylsilylimidazole of the formula 45 50 wherein Y is an alkyl group, preferably methyl, with ar irylalkylhalogenide of the formula 55 wherein $R_{\rm 3}, R_{\rm 5}, R_{\rm 8}, R_{\rm 7}, R_{\rm 8}$ and $R_{\rm 9}$ are as defined in claim 1 and Hal is a halogen atom, in the presence of of a

GB 2 101 114 A

Lewis acid.

20

70

55

25

72. A process for the preparation of a compound as claimed in Claim 1, which comprise reading a starting material of the formula

10 wherein R_b, R_b, R_b, R_b, R_b and n are as defined in claim 1 and R_{1b}, R_{1b}, R_{bl} and R_{1b}, which can be the same or different, are each hydrogen, hydroxy, mercapto, halogen, amino, = 0 = alkyl of 1 to 7 carbon stores or

wherein R is an alkyl; or wherein R_{12} and R_{14} can be combined to form a lasto group, or R_{13} and R_{14} can be combined to form a lasto group, or both R_{12} and R_{14} and R_{13} and R_{15} can simultaneously form lasto groups, with a respect capable of converting said starting material to the corresponding imidazole.

with a reagent capable of converting said starting materials of the contract of the comprises reacting.

73. A process for the preparation of a compound as claimed in Claim 1 which comprises reacting formamide with a compound of the formula.

wherein R₅, R₆, R₇, R₈ and n are as defined in Claim 1 and Q is

5 (wherein Hal is a halogen atom, R is substituted or unsubstituted alkyl, aralkyl or anyl group), provided that a) when Q is

45 the reaction with formamide is followed by treatment of the intermediate product with acid; and

50 b) when
$$q \mapsto \frac{y}{y} = \min_{k_2} \frac{y}{k_2}$$
.

the reaction with formamide is followed by hydrogenation of the intermediate product.

74. A process for the preparation of a compound as claimed in Claim 1 which comprises hydrolysing a
55 compound of the formula

ED) GB 2 101 114 A $\mathfrak{S}\mathfrak{D}$ where Y is the envisity residue determined by the fermula (ii) and (ii), A is on easy group of I is I excens etoms on any recital of 6 to 10 carbon etoms. 78. A process for the preparation of a compound as stamed in Carm 1 which comprises the compound of clumed cit to brucemed S 3 co 100 ٤, 15 ක්ෂාත්ත V හි යා රෝකාන් හි **වාර්ත 74. බ**් හි හා හැනි හා වෙනු ගත් බී. ය හා හැනි ඉතර 78. A present for the preparation of a compound as alternation Quem 1 कर्याच्या कर कर कराया न कर one of the faregoing Examples. 77. A compound to defined in empans el Clama 1 to 60 ක්කා ලාදයක් හැ ම ගතනය යා ක්කාය ය emperes of Claims 64 to 75. CI 78. A substituted imidentity of the general farmula 3 23 \square 30 B wherein R₁ is H, an alkyl of 1 to 4 carbon atoms or - CM₂CM; R₂ is M or CH₁; R₁ is 9 40 **4**3 45 or R₃ and R₄ together represent = CH₂, = CH - CH₃, ಣ 50 =CH-CH2-CH3, =C CH3, or =CH-CH2CH2CH3; X is R. 1 R10 Ra -CH-CH- or -C=C-; R₀, R₀ and R₂, which can be the same or different, and H, $-CH_0$, $-CM_7CM_0$. 62 55 halogen, OH or - OCH3 or Ro is hydrogen and Ro and R, together form an O CH, O bridge between two adjacent carbon atoms in the phenyl group; —CHR $_8$ - is -CH $_2$ - , CH(CH $_3$) ,

-CH(-CH2CH3)-, -CH(-CH2CH2CH3)-, -CH(-CH-CH3)-,

-CH(-CH2CH2CH2CH3)- or C=CH2, C=CH-CH3, C=CH-CH2CH1,

60

- 20 A 600 003 8 60 83 _ೆಯಿ² -OH, OH, OH, : -OH - ON, -OH, OH, OH, OH, OH, OH, H $_{\odot}$ OH, H $_{\odot}$ IS H. - CH $_{\Delta}$ **P O**₩_® -OHS -CHIONS - CHIONIONS OH-OHS CI - ON ON ON ON n is 0 to 4, provided that Ro and Ro or Ro, and Ro are not simultaneously hydrogen, and is recess 10 pharmacetically acceptable acid addition salts, for use in therapy as an entity percessive again. antithrombotic agent, antimyeotic or antilungal eyest. 79. A pharmaceutical composition comprising on imidates democrac to defined a Com 13 ct o non-toxic pharmaceutically acceptable seed edulion salt thereof and a compatible charmaceutically acceptable carrier therefor. 80. A pharmaceutical compection computing on unidates derivativo as element in any cro dici ..22 Claims 1 to 63 or 77 or a non-term phermerouseally acceptable and addition sell thereof and a company pharmaceutically assertable carrier therefor.

(2) جو با مردی النصا (مدری) و المردی مردی و المردی و الم